

SHORT  
COMMUNICATIONS

## Oxidation of 1-Oxa-5-azaspiro[5.5]undecane with Cyclohexanone

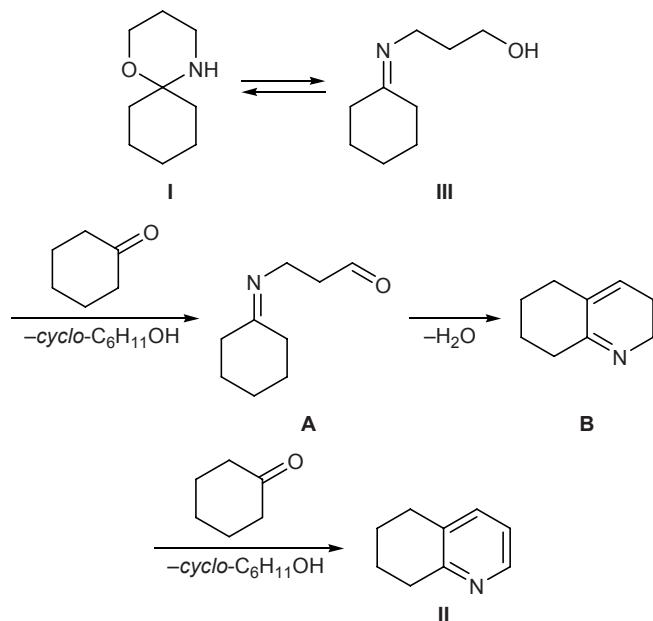
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We previously reported that oxidation of five-membered cyclic N,O-acetals (oxazolidines) with carbonyl compounds leads to the formation of pyrroles [1]. In the present communication we describe the oxidation of 1-oxa-5-azaspiro[5.5]undecane (**I**) with cyclohexanone to produce 5,6,7,8-tetrahydroquinoline (**II**) as an example of synthesis of pyridine derivatives via oxidation of perhydro-1,3-oxazines having no substituent on the nitrogen atom with carbonyl compounds.



Presumably, as in the oxidation of oxazolidines with carbonyl compounds, initial oxidation of imino alcohol **III** (an open-chain tautomer of perhydro-1,3-

oxazine **I**) gives imino aldehyde **A** which undergoes intramolecular condensation (like crotonization) to 2,3,5,6,7,8-hexahydroquinoline **B**, and the latter is oxidized with the second cyclohexanone molecule to quinoline **II**.

A mixture of 31 g (0.2 mol) of compound **I**, 39.2 g (0.4 mol) of cyclohexanone, and 4.48 g (0.08 mol) of powdered potassium hydroxide was heated for 5 h under reflux. The mixture was cooled, 25.2 ml (0.21 mol) of 36% hydrochloric acid was added, and the aqueous phase was separated, extracted with diethyl ether ( $3 \times 50$  ml), made alkaline by addition of potassium hydroxide, and extracted with benzene. The benzene extract was subjected to distillation to isolate 4.2 g (16%) of 5,6,7,8-tetrahydroquinoline **II** with bp 89–93°C (10 mm),  $d_4^{20} = 1.0314$ ,  $n_D^{20} = 1.5428$ ; published data [2]: bp 92–95°C (12 mm),  $d_4^{20} = 1.0304$ ,  $n_D^{20} = 1.5435$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm ( $J$ , Hz): 1.74 m (4H, 6-H, 7-H), 2.68 t (2H, 5-H), 2.80 t (2H, 8-H,  $J_{5,6} = 6.0$ ,  $J_{7,8} = 6.0$ ), 7.0 d.d (1H, 3-H,  $J_{3,4} = 8.0$ ), 7.33 t (1H, 4-H), 8.24 d (1H, 2-H,  $J_{2,3} = 5.0$ ). Found, %: C 81.33; H 8.26; N 10.41.  $C_9H_{11}N$ . Calculated, %: C 81.16; H 8.32; N 10.52.

## REFERENCES

1. Kukharev, B.F., Stankevich, V.K., and Kukhareva, V.A., *Khim. Geterotsikl. Soedin.*, 1984, p. 700.
2. Svoistva organicheskikh soedinenii. Spravochnik (Properties of Organic Compounds), Potekhin, A.A., Ed., Leningrad: Khimiya, 1984.